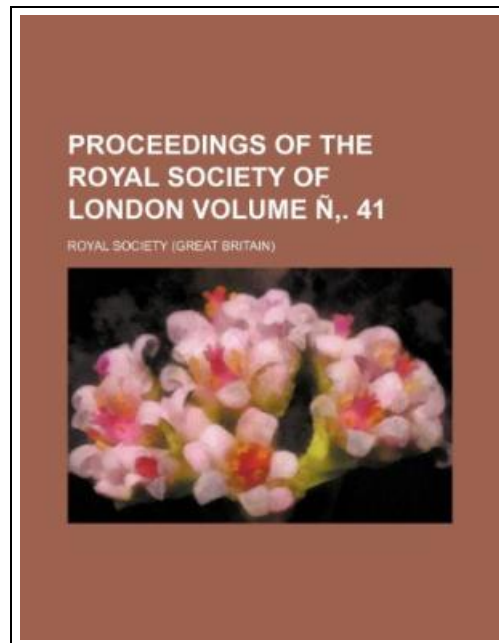


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1887 Excerpt: .easily effected as we effect the determination of the specific gravity of a body, there are on the other hand sufficient reasons to recommend its use as in general of more physical value and interest than the much used Specific gravity. There are cases indeed where specific gravity, as it is possible to obtain it, is misleading, and where specific heat gives at once valuable information on the probable chemical nature of the substance. Such cases would arise with bodies of loose vesicular or hollow structure. No misleading variations need be introduced into the thermal constant by mere conditions of volume. The method of condensation permits of the determination of this constant with very little experimental difficulty. On the whole I have to thank Professor Sollas for the loan of useful specimens from the Museum of Trinity College. + In November, 1883, I suggested this use of calorimetry to the Experimental Science Association, Trinity College.-- On the Determination of Minerals by their Specific Heat. process is involved with no more liability to error than the process of weighing a body in air and in water. If, however, accuracy--greater than 1 per cent.--be desired, the operation takes considerably more time, as it is necessary, in securing tx accurately, to leave thermometer and specimen a sufficient time together in the calorimeter. The simplicity of the method is perhaps best inferred from the fact that of over 130 experiments on minerals there were but three spoiled--due to mishaps;...



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