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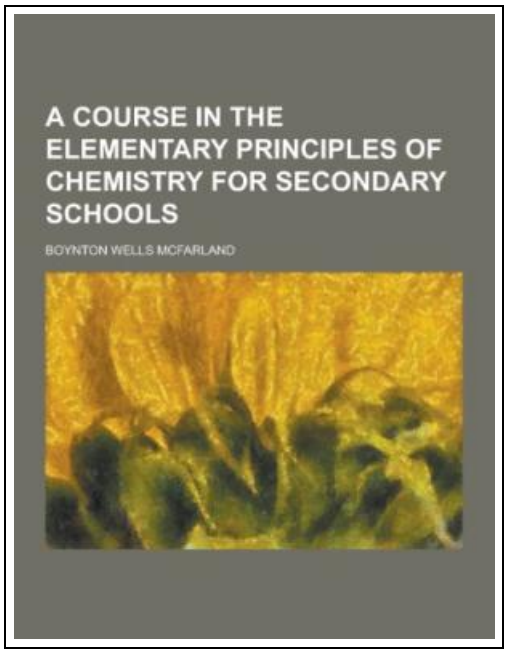

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A COURSE IN THE ELEMENTARY PRINCIPLES OF CHEMISTRY FOR SECONDARY SCHOOLS

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Theclassics.Us, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1907 edition. Excerpt: .is that which has weight. 2. An element is a substance that contains only one kind of matter. There are about seventy-five elements, of which only about twenty-five are common. Examples of common elements are: iron, copper, lead, tin, gold, and silver. 167-168. 3. A symbol of an element is a sign that stands for that substance. It ordinarily consists of one or two characteristic letters of the English or Latin name. 4. Symbols and names of some common elements, in groups. 167. Not an element, but a compound of N and H that acts like one. 5. Properties of matter. Those qualities which are peculiar to it and characteristic of it. An exact description of any kind of matter involves telling what it looks like, smells like, tastes like, how it behaves at different temperatures and when mixed with other substances. 6. All matter is supposed to be made up of a very large number of very small particles called molecules, and these molecules to be made up of still smaller particles called atoms. 7. A molecule is the smallest particle of matter that can exist and still retain all the properties of that kind of matter. If the substance is an element, the molecule will consist of one or more atoms of that element; if the substance is a chemical compound, the molecule will contain at least one atom of every element contained in the substance. 8. An atom is the smallest...

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