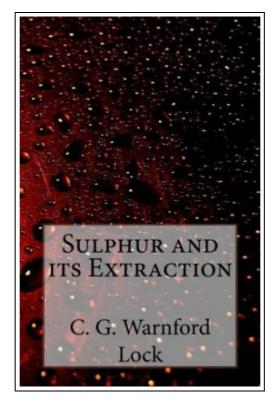
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Createspace Independent Publishing Platform, United States, 2016. Paperback. Condition: New. Language: English. Brand New Book ***** Print on Demand *****THE following notes relate exclusively to native sulphur (brimstone). Though the amount of sulphur annually rained in the form of sulphides of various metals (e. g., iron and copper pyrites, galena, blende, etc.) probably far exceeds that obtained in the uncombined state, still, the separation of the sulphur in an inoxidized condition from such compounds is never attempted, for the simple reasons that, in the processes for extracting the several metals from their ores, the first step necessary is the elimination of the combined sulphur, which is most easily effected by a roasting or oxidizing operation, whereby the sulphur is at once converted into sulphurous acid, itself a valuable commodity, and, moreover, capable of being readily oxidized one step further to form sulphuric acid, the chief purpose for which sulphur is consumed. There are two mines of sulphur worked in Austria-Hungary, one not far from Cracow, and the other at Radoboi in Croatia; both deposits are of considerable extent, but the annual yield is insignificant. The whole district around Mount Budos, in Transylvania, is rich in sulphur. Some thirty or more diggings have been undertaken in a circuit of eighteen miles, but the area covered by the deposits is more than three times this size. The sulphur occurs in unequal strata one to nine inches thick, beneath one to three feet of mold. The soil is everywhere saturated with sulphur, and in this permeated earth pieces of the pure mineral are found. The whole is the result of living solfataric action, and the accumulation will continue to grow as long as that action survives. Samples of the impregnated earth, taken over an area of 16,000,000 square fathoms, yielded from forty-one...

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